

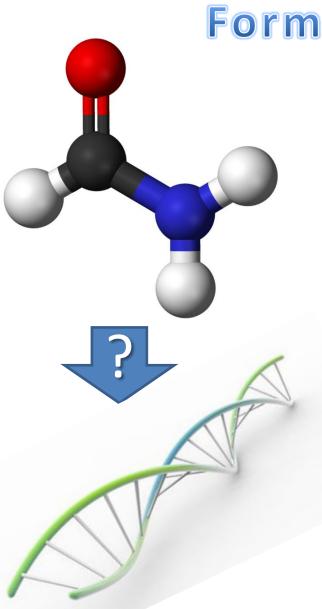


AKADEMIE VĚD ČESKÉ REPUBLIKY Czech Academy of Sciences

# Role of Formamide in Chemical Evolution of Biomolecules

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## Formamide Molecule

#### "Molecule with a prebiotic potential"

#### WHY?

- This molecule contains all the esential heteroatoms important for life.
- The structure represents the simplest model of a peptide bond.
- It is sufficiently reactive upon UV, shock waves, protons, discharges (details later).
- Without presence of radiation or other energy sources, it is a stable liquid.

#### **First Steps**

**1956** Hellmut Bredereckef **Neue Pyrimidin-Synthese aus β-Dicarbonyl-Verbindungen und** Formamid

Formamid-Reaktionen: Eine Neue Pyrimidin-Synthese

One Step Synthesis of Purine Ring from Formamide

2001 Raffaele Saladino

1972

Hiroshi Yamada

A possible prebiotic synthesis of purine, adenine, cytosine, and 4(3H)-pyrimidinone from formamide: Implications for the origin of life

Proposed Scenario:

Formamide is related to HCN chemistry.

It might be produced by HCN hydrolysis in water, concentrated by evaporation and then react being heated, exposed to UV ( ..... or to other sources of sufficiently high energy).

### Formamide Heating

#### 160°C + selected catalysts

Mechanism is likely (heat flux on early Earth was sufficient).

First experiments: Silica, Alumina: purine, adenine, cytosine, pyrimidinone Current Experiments: Meteorites: nucleic bases, sugars, some aminoacids (Saladino) No catalyst: All the canonical nucleic bases (Ferus, Civiš)

#### Formamide + UV Radiation

#### 130°C and 254 nm (Barks et al.)

Mechanism is maybe not so likely (heavy clouds and dust in the atmosphere).

Guanine, Adenine, and Hypoxanthine



#### Formamide + Cosmic radiation

#### Simulation of Solar Wind by a Proton Flux

In presence of meteorites: Formation of all the canonical nucleic bases, sugars and selected aminoacids (Saladino).

Extraterrestrial source of biomolecules (chemical panspermia).

#### Formamide + Impact Shock Waves

Simulation using a large laser facility

Formamide + Lighting Discharges

Likely.

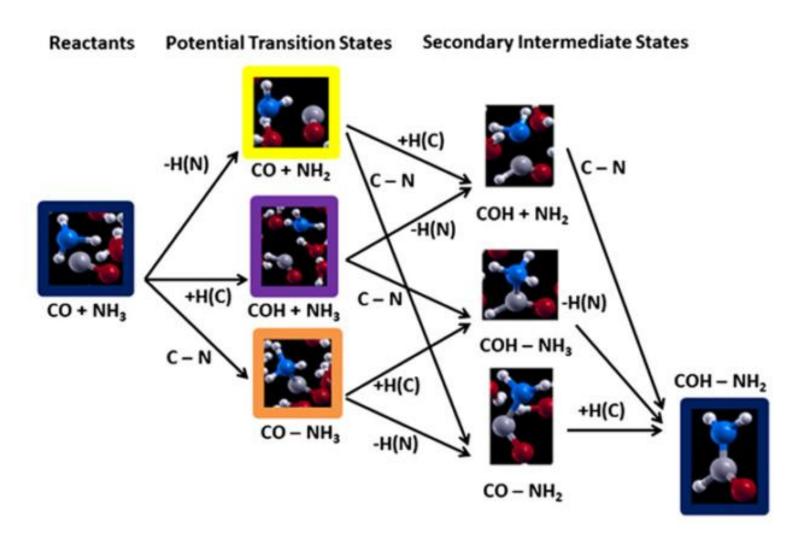
BUT:

Still not well explored.

ormamide is an intermediate of aminoacids formation (Saitta).

### **Formation of Aminoacids**

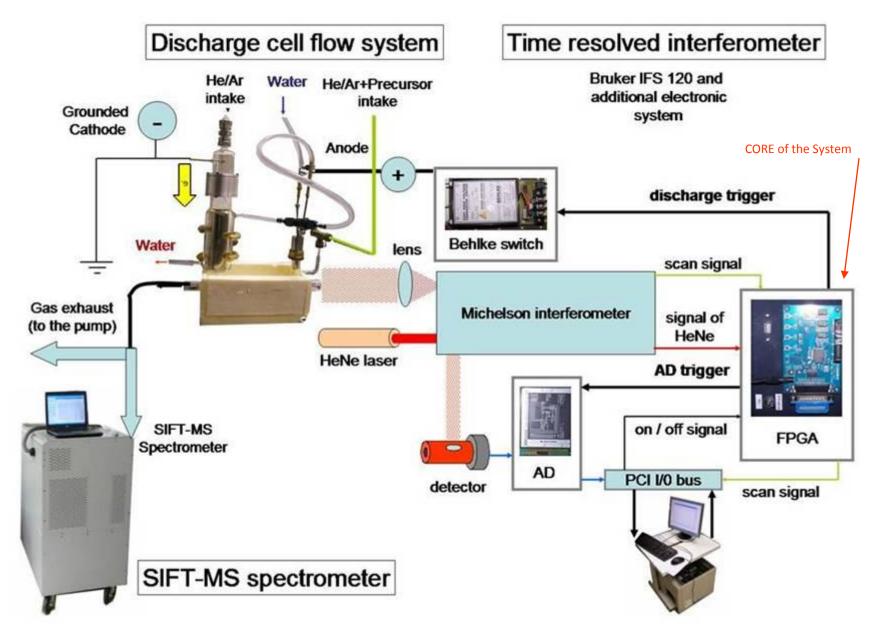
Theoretical calculations on Miller – like Experiments (Saitta et al.) Formamide is an intermediate of aminoacids formation

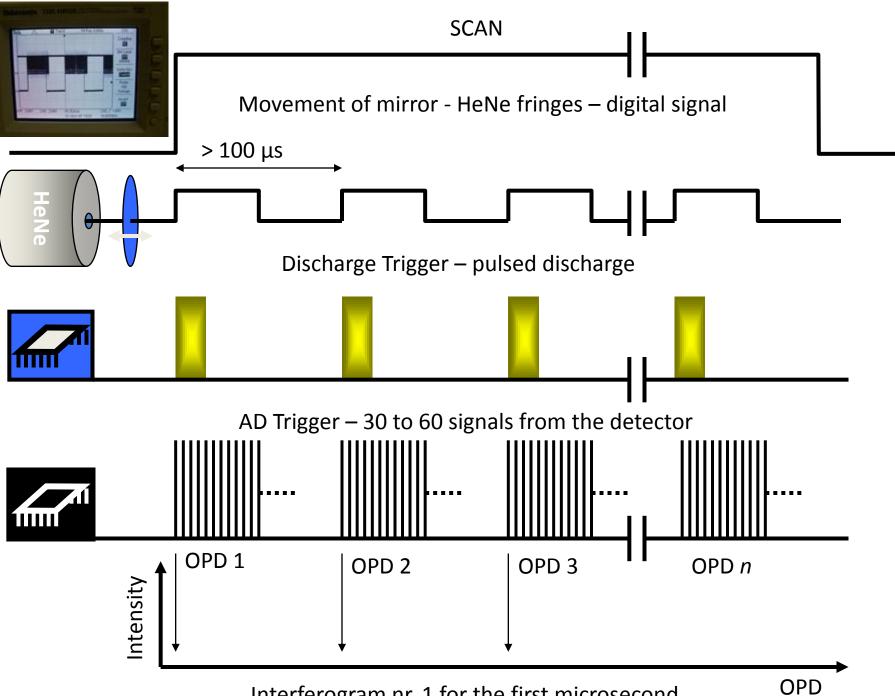


### Time-Resolved Spectroscopy

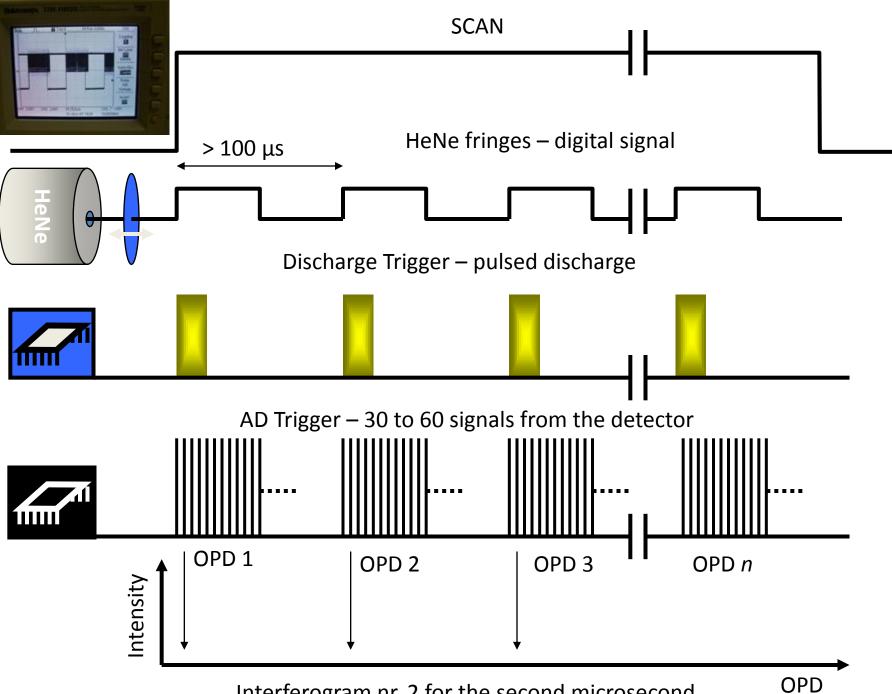
Exploration of Formamide Chemistry in Discharges

# **Electronic set-up**

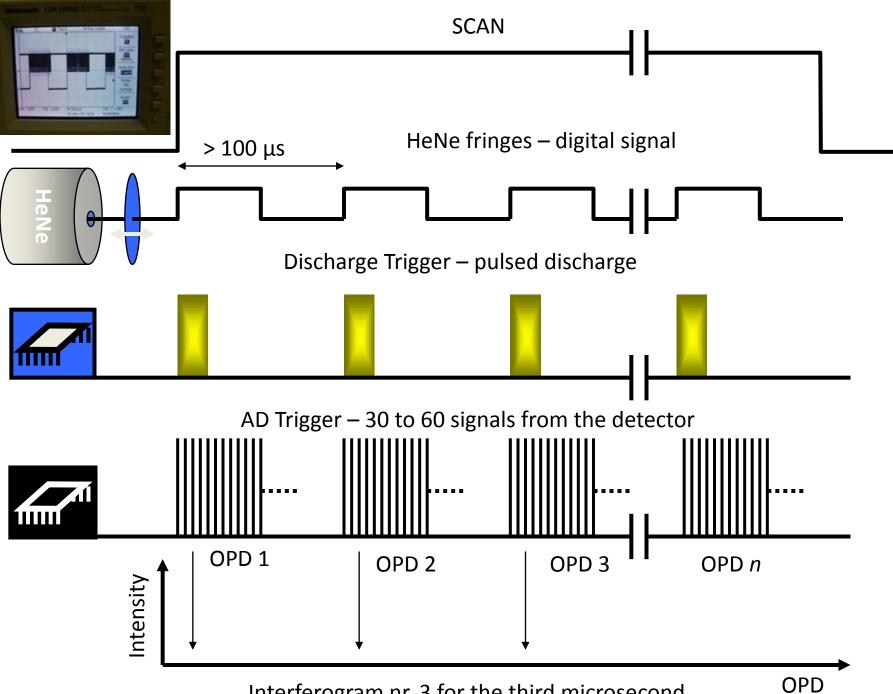




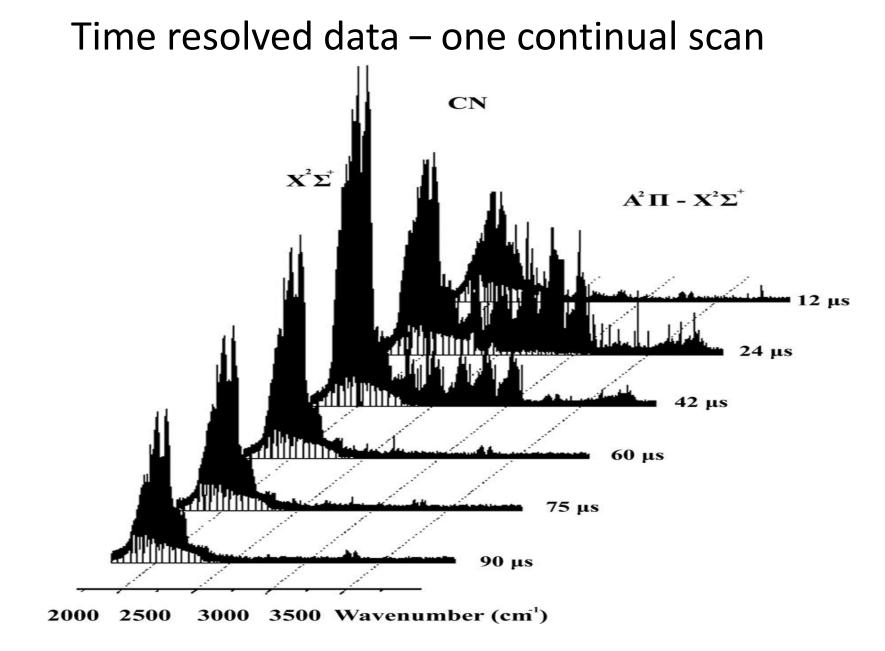
Interferogram nr. 1 for the first microsecond



Interferogram nr. 2 for the second microsecond

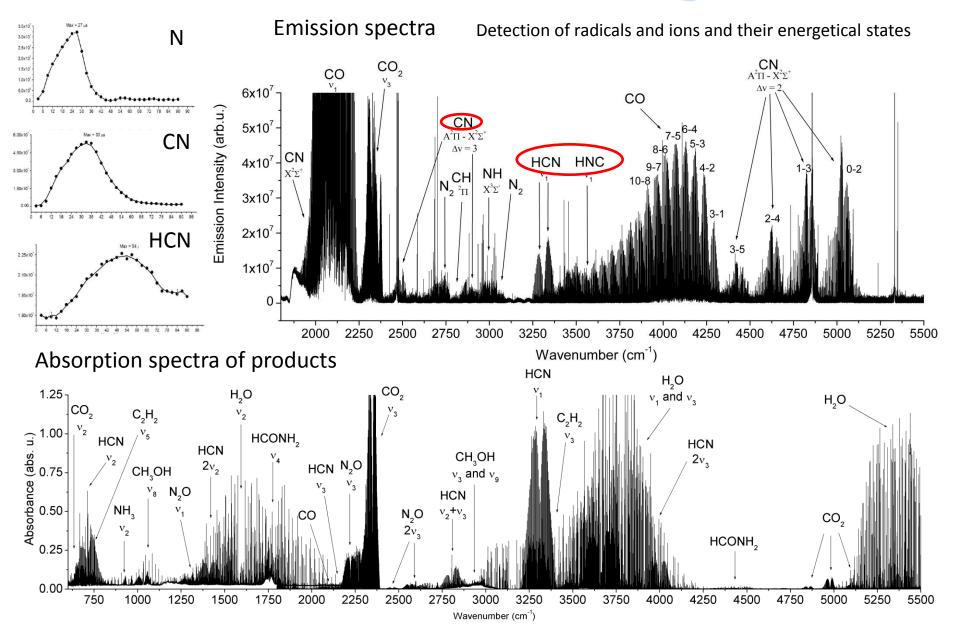


Interferogram nr. 3 for the third microsecond

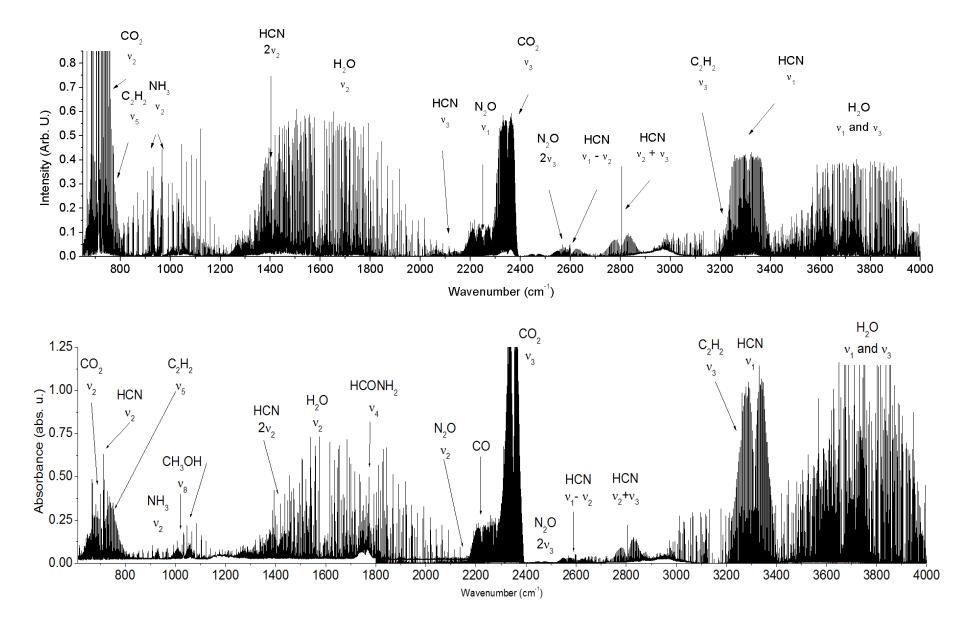


The time-resolved emission FT spectrum from a pulsed discharge in a  $(CN)_2$  and He mixture. The discharge pulse duration was 20µs. The 30 time-resolved spectra were collected from t = 0–90 µs with a step of 3 µs. The spectra of  $C_2H_2$  and  $C_2$  were observed at 3300 and 3600 cm<sup>-1</sup>.

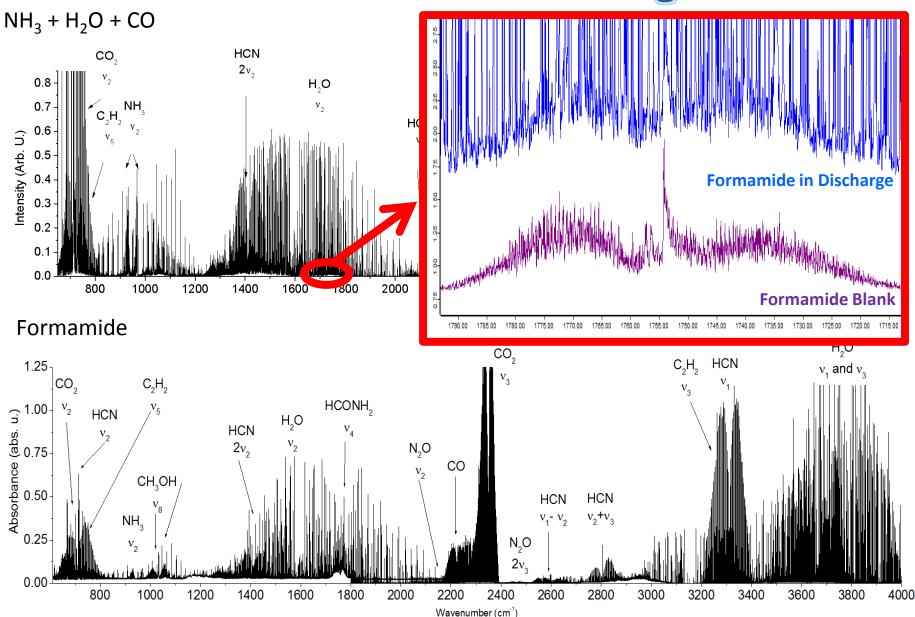
### **Formamide Discharge**



#### **Formamide Discharge**



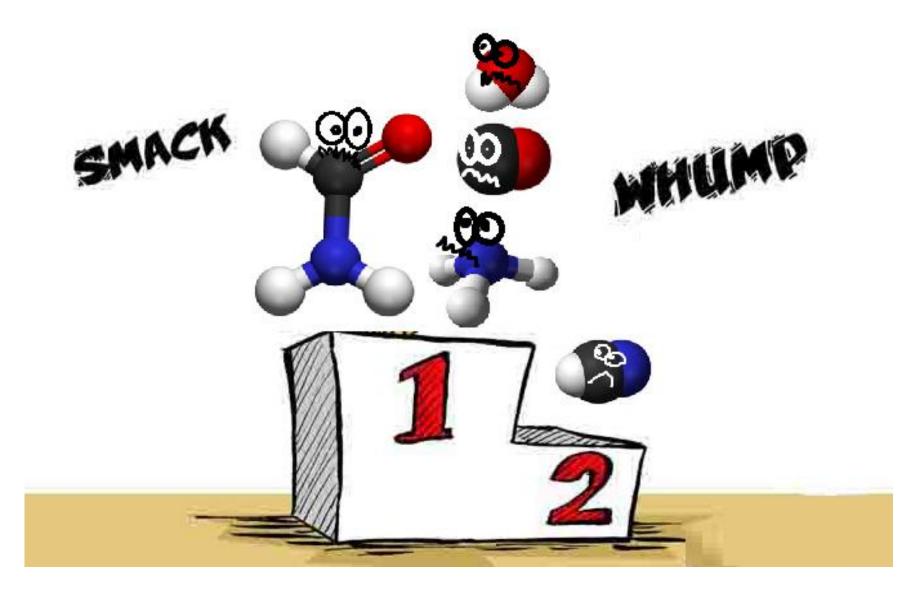
### **Formamide Discharge**



#### THE OLD CHICKEN AND EGG PROBLEM ....



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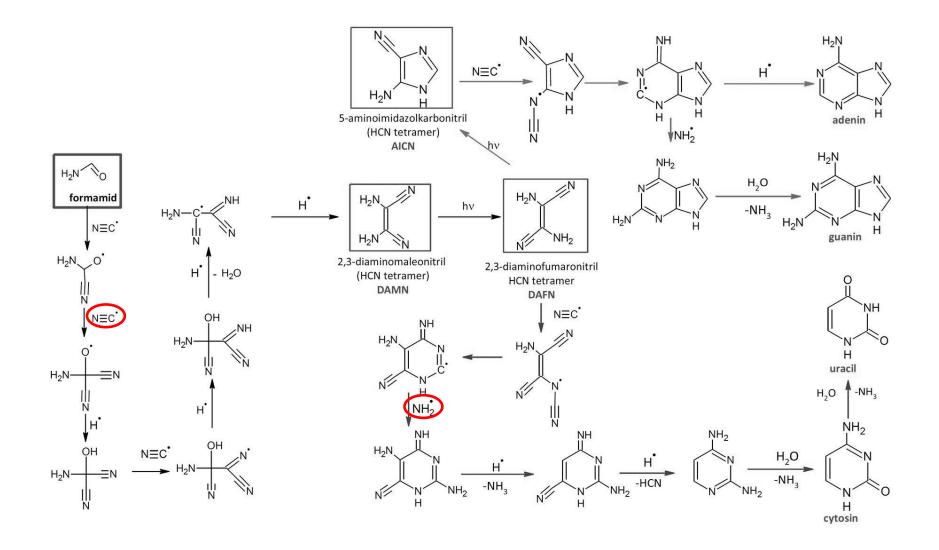
Mutual Conversion in Complicated Reaction Chains leads to Formation of Biomolecules.

Formamide Plays an Important Role of Reactive Substrate (Urey-Miller) or Parent Compound (Saladino).

Fundamental role is played by CN, CO, NH<sub>2</sub>

Т

#### **Cyano Radical Mechanism**



### **Cyanide Mechanism**

